JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT TEST -2 EXAMINATION- Oct 2017

B.Tech 7th Semester

| | B.Tech / Semester | |
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| COURSE CODE: 10B13CE742 | | MAX. MARKS: 25 |
| C | OURSE NAME: Air Pollution Monitoring and Control | |
| COURSE CREDITS: 3 MAX. TIME: 1Hr 30 M | | MAX. TIME: 1Hr 30 Min |
| No | ote: All questions are compulsory. Carrying of mobile phone | during examinations will be |
| tre | rated as case of unfair means. | |
| 1. | Determine the effective height of a stack, given the following da | ta: |
| | (a) Physical stack is 170 m tall with a 1.25 m inside diameter | |
| | (b) Wind velocity is 5.17 m/s | |
| | (c) Air temperature is 18 °C | |
| | (d) Atmospheric pressure is 1000 millibars | |
| | (e) Stack gas velocity is 8.75 m/s | |
| | (f) Stack gas temperature is 128 °C | (4) |
| 2. | A parcel of dry air rising over a grass fire has a temperature of 60 °C at 10 m. Assuming a | |
| | dry adiabatic lapse rate, determine the temperature at 200 m. | (2) |
| 3. | Write short notes on dry adiabatic lapse rate and wet adiabatic la | pse rate. (2) |
| 4. | Define Air pollution and air pollutants. Discuss and state various sources of air pollution. (2) | |
| 5. | The exhaust gas from an automobile contains 1.0 % by volume of CO. What is the | |
| | concentration of CO in mg/ m³ at 25°C and 1 atm pressure? | (2) |
| 6. | If SO ₂ , CO and NO ₂ are found out to be 1000 ppm, 600 ppm and 200 ppm respectively in a | |
| | air quality survey, What are concentrations in µg/m3? | (3) |
| 7. | Write a detailed note on - 'Structure of the atmosphere'. | (3) |
| 8. | Discuss general effects of air pollution. | (2) |
| 9. | process, the Plant has the following information. Sulphur in coal: 0.5% (weight/weight) Temperature in stack: 125°C Pressure in stack: 1.5 atm Stack exit velocity: 15m/s Diameter of stack: 10m | |
| | Estimate SO ₂ concentration in kg/day and ppm. | (5) |