

JAYPEE UNIVERSITY OF INFORMATION TECHNOLOGY, WAKNAGHAT

TEST -2 EXAMINATION- 2024

B.Tech – III<sup>rd</sup> Semester (BE)

COURSE CODE (CREDITS):18B11CE311(3)

MAX. MARKS: 25

COURSE NAME: Chemistry

COURSE INSTRUCTORS: Dr. Poonam Sharma

MAX. TIME: 1 Hour 30 Minutes

*Note: (a) All questions are compulsory.*

*(b) The candidate is allowed to make Suitable numeric assumptions wherever required for solving problems*

Q.No	Question	CO	Marks						
Q1(a)	Elucidate surface tension and interface tension.	CO-2	2						
(b)	Elaborate the applications of Colloids.	CO-2	3						
Q2(a)	Explain the solubility and various factors affecting it.	CO-2	2						
(b)	Discuss various remedial measures to control Ozone depletion.	CO-6	3						
Q3(a)	How exhaust vehicular emission can be reduced by correct Air/fuel ratio?	CO-6	3						
(b)	The density and associated percent crystallinity for two nylon 6,6 materials are as follows: <table style="margin-left: 40px; border: none;"> <tr> <td style="padding-right: 40px;"><math>\rho</math> (g/cm<sup>3</sup>)</td> <td>crystallinity (%)</td> </tr> <tr> <td>1.188</td> <td>67.3</td> </tr> <tr> <td>1.152</td> <td>43.7</td> </tr> </table> (a) Compute the densities of totally crystalline and totally amorphous nylon 6,6. (b) Determine the density of a specimen having 55.4% crystallinity.	$\rho$ (g/cm <sup>3</sup> )	crystallinity (%)	1.188	67.3	1.152	43.7	CO-4	5
$\rho$ (g/cm <sup>3</sup> )	crystallinity (%)								
1.188	67.3								
1.152	43.7								
Q4(a)	Calculate the normality of NaOH solution formed by dissolving 0.2 gm NaOH to make 250ml solution	CO-2	3						
(b).	If polymer sample has population as: 10 molecules of molecular mass each = 5000 20 molecules of molecular mass each = 7500 20 molecules of molecular mass each = 10000 25 molecules of molecular mass each = 15000 20 molecules of molecular mass each = 20000 5 molecules of molecular mass each = 25000 Calculate its number average and weight average molecule mass of polymer and PDI	CO-4	4						